

- (3) A. Itaya, K. Okamoto, and S. Kusabayashi, *Bull. Chem. Soc. Jpn.*, **49**, 2082 (1976).
 (4) A. Itaya, K. Okamoto, and S. Kusabayashi, *Bull. Chem. Soc. Jpn.*, **50**, 22 (1977).
 (5) C. E. Hoyle and J. E. Guillet, *Macromolecules*, **11**, 221 (1978).
 (6) C. E. Hoyle and J. E. Guillet, *Macromolecules*, **12**, 956 (1979).
 (7) U. Lachish, R. W. Anderson, and D. J. Williams, *Macromolecules*, **13**, 1143 (1980).
 (8) U. Lachish, D. J. Williams, and R. W. Anderson, *Chem. Phys. Lett.*, **65**, 574 (1979).
 (9) G. E. Johnson, *J. Chem. Phys.*, **61**, 3002 (1974).
 (10) W. Klöpffer and D. Fischer, *J. Polym. Sci., Part C*, **40**, 43 (1973).
 (11) A. Itaya, K. Okamoto, and S. Kusabayashi, *Bull. Chem. Soc. Jpn.*, **49**, 2037 (1976).
 (12) R. D. Burkhart and R. G. Aviles, *J. Phys. Chem.*, **83**, 1897 (1979).
 (13) W. Klöpffer, *J. Chem. Phys.*, **50**, 2337 (1969).
 (14) F. Schneider, *Z. Naturforsch., A*, **24**, 863 (1969).
 (15) R. D. Burkhart and R. G. Aviles, *Macromolecules*, **12**, 1073 (1979).
 (16) W. Klöpffer, N. J. Turro, M. F. Chow, and Y. Noguchi, *Chem. Phys. Lett.*, **54**, 457 (1978).
 (17) G. Rippen and W. Klöpffer, *Ber. Bunsenges. Phys. Chem.*, **83**, 437 (1979).
 (18) D. J. Freed and Larry R. Faulkner, *J. Am. Chem. Soc.*, **94**, 4792 (1972).

Uri Lachish and David J. Williams*

Xerox Corporation
 Webster Research Center
 Webster, New York 14580
 Received February 1, 1980

CORRECTIONS

H. R. Allcock,* T. L. Evans, and D. B. Patterson: Reaction of Poly(difluorophosphazene) with Phenyl-lithium. Volume 13, Number 2, March–April 1980, page 203.

Three errors appear in Table I. The fourth column heading should read % PF₂, not % PR₂. The entries in the third and fourth columns of the last row should be interchanged.

G. A. Senich and W. J. MacKnight*: Fourier Transform Infrared Thermal Analysis of a Segmented Polyurethane. Volume 13, Number 1, January–February 1980, p 106.

The numerical values given in Table I (p 109) for the amount of hydrogen-bonded N–H and C=O are not percentages as indicated but fractions of the unit total and each is uncertain by a factor of 2% of the given value. A corrected version of Table I appears below.

Table I
 Changes in the Degree of
 Hydrogen Bonding with Temperature

temp, °C	fraction bonded N–H ^a	fraction bonded C=O ^a
29	0.875	0.600
42	0.845	0.585
53	0.790	0.570
63	0.770	0.560
74	0.760	0.545
88	0.695	0.520
98	0.650	0.500
114	0.605	0.470
126	0.565	0.450
136	0.550	0.430
151	0.500	0.410

^a Accurate to ± 2%.